## Chem 41c Quiz 1

Stoltz, Spring 2010 April 9, 2010

You have 25 min to take this quiz. It is closed note, closed book, and no collaboration is allowed. Please do not discuss the quiz with anyone until you receive it back graded. Place a box around your answers. There is no partial credit.

Predict the products (if any) of the following reactions: (5 points each)

1.

2.

3. In the following complexes, what is the formal oxidation state of the metal, the  $d^n$  description, and the electron count? Feel free to use the periodic table in the room (no partial credit). (5 points each)

Pd(PPh <sub>3</sub> ) <sub>4</sub>	Pd(0) or Pd <sup>0</sup>	d <sup>10</sup>	18 electron complex
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b.

$$\begin{array}{c} \text{Rh(I)} \\ \text{RhCI(PPh}_3)_3 & \text{or} \quad \text{d}^8 \qquad \text{16 electron complex} \\ \text{Rh}^{1+} \end{array}$$

Bonus (5 points)
Predict the product of the following reaction. Be sure to indicate stereochemistry in your proposed product.

